



QUANTITATIVE OPTICAL DIAGNOSTICS OF PLASMAS

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Course Overview

Underlying science: atomic and molecular spectroscopy





- Introduction to fundamentals of atomic and molecular spectroscopy
- Quantitative emission and absorption spectroscopy
- Laser induced fluorescence (LIF) and Two-photon Absorption laser induced fluorescence (TALIF)
- Laser Scattering techniques (Rayleigh, Raman, Thomson)
- Non-linear techniques (CARS, E-CARS, E-FISH)

Fundamentals of Atomic and Molecular Spectroscopy

Additional Information

Wolfgang Demtröder

Graduate Texts in Physics

Atoms, Molecules and Photons

An Introduction to Atomic-, Molecularand Quantum Physics

Third Edition

Slides are not complete, a few textbooks can be useful

Deringer

Electronic Transitions of the Hydrogen Atom

Hydrogen atom energy levels





Two-level System: Optical Transitions



 $\left(\frac{dn_2}{dt}\right)_{spontaneous} = -A_{21}n_2$

E_i: energy of level I

n_i: population of level *i*

A₂₁: Einstein coefficient for spontaneous emission (transition probability)

 $B_{12}\rho(v)$, $B_{21}\rho(v)$: transition probability for absorption, stimulated emission

ρ(v): spectral density function

$$\left(\frac{dn_1}{dt}\right)_{absorption} = -B_{12}\varrho(\nu)n_1$$

Emission Intensity





Roettgen et al, Plasma Sources Sci. Technol. 25 (2016) 055009

Local emissivity

$$e_{\lambda} = \frac{1}{4\pi} h v A_{21} N_2 \phi \left(\lambda \right)$$

Wavelength-dependent measured intensity

$$I_{\lambda} = \int_{l} e_{\lambda}(l) \, \mathrm{d}l = \frac{1}{4\pi} Lh \nu A_{21} N_{2} \phi(\lambda).$$

Uniform volumetric absorbing medium

 $\Phi(v)$: spectral line profile

- Information about excited states can be accessed
- $\Phi(v)$ cannot be Dirac's function because of Heisenberg's uncertainty principle \rightarrow Natural broadening

Line Broadening

Atomic line broadening due to:

- Doppler broadening (thermal broadening, temperature effect) $\Delta \lambda_{Doppler} \propto \sqrt{T_{g}}$
- Van der Waals broadening (emitter perturbed by atoms and molecules)
- Stark broadening
- Instrumental broadening / apparatus function (finite resolution of the spectrometer)

for high resolution spectrometer, width of spectral lines provides information on gas temperatures!

Czerny-Turner spectrometer



A. Incoming light sourceB. Entrance slitC. Collimating mirrorD. Grating on turretE. Focusing mirrorF. Exit slitG. Single channel detector

$$\Delta\lambda_{vdW} \propto \frac{p}{T_g^{0.7}}$$

Diatomic Molecules



Molecular Spectra



Rotational Temperature Effects





$$I_{JJ'} \propto (2J+1)A_{JJ'} \nu \exp(\frac{-E_J}{kT_{rot}}) \longrightarrow \ln(\frac{I}{(2J+1)A_{JJ}\nu}) = (\frac{-E_J}{kT_{rot}}) + cst$$

Quantitative Emission and Absorption Spectroscopy

Optical Emission Spectroscopy (OES)



Laux et al, Plasma Sources Sci. Technol. 12 (2003) 125-138

David Pai, PhD Thesis Ecole Centrale Paris (2008)

- Passive diagnostic
- Line-of-sight integrated
- High temporal resolution and 1D measurements achievable
- Primarily gives information about species in their excited states
- Can be used to measure: temperatures, neutral species densities, electron densities and electric field vectors

Typical Emission Spectra in Plasmas



- $N_2(C-B)$, OH(A-X), $N_2^+(B-X) \rightarrow Rotational$ temperatures measurements $\rightarrow Gas$ temperatures measurements
- Atomic hydrogen lines → Stark broadening → Electron densities measurements
- Atomic hydrogen lines → Stark shifting → Electric field vector measurements

Excited States Absolute Densities Measurements



- Densities of $N_2(B)$ and $N_2(C)$ were obtained from $N_2(B-C)$ and $N_2(B-A)$ spectra
- Absolute densities obtained using standard quartz tungsten halogen (QTH) lamps

Electron Density Measurements



- Balmer β line is very reliable for n_e measurements as its FWHM is only weakly dependent on T_e and on ions dynamics
- A deconvolution is necessary to extract the part of the broadening due to Stark broadening

Rotational/Gas Temperature Measurements using N₂(C-B)



2-m focal length spectrometer from McPherson



CCD detector from Andor Technologies



- High resolution (4 pm) rotational spectrum of the N₂ (C-B; 0-0) vibrational band in a glow discharge
- The 4 pm spectral resolution is obtained using a 2-m focal length spectrometer

Bruggeman et al, Plasma Sources Sci. Technol. 21 (2012) 035019)

Absorption Spectroscopy



• Active diagnostic

• Line-of-sight integrated

- High temporal resolution
- Primarily gives information about species in their ground state
- Can be used to measure: temperatures and neutral species densities

Stable light sources required

Light sources in the UV/Vis and Mid-IR region are of great interest!

Absolute Densities Measurements



 $n_i = n_{tot} f_B(T)$

Radial Quantities Determination: Abel Inversion



Radial quantities can be obtained

Absolute Measurements of Low Density Species

500

0

1000

Time (µs)

1500

White multi-pass cell



Cavity Ring-down Spectroscopy



2000

- Long absorption lengths: tens of meters to kilomet
- Ultrasensitive measurements: trace species densities measurements down to parts per billion

Laser Induced Fluorescence Diagnostics

Laser-induced Fluorescence (LIF)

Principle



Planar LIF system for combustion



- Active diagnostic
- High Sensitivity/selectivity
- 2D imaging capabilities
- Enable measurements of temperatures and densities
- Spontaneous technique

Molecular species: NO, OH, CH, CN, O₂, C₂

Atomic species: N, O, H

NO LIF Setup



Van Gessel et al J. Phys. D: Appl. Phys. 46 (2013) 095201

- Pump laser (harmonics of Nd:YAG)
- Dye laser
- Harmonic crystal

OH LIF



rotational and vibrational energy transfer (RET, VET) and electronic quenching

Temperature Measurements: Two LIF Approaches











Absolute Densities Measurements: 2-level LIF



- A model is required to relate the measure LIF signal to the ground state density
- The model contains multiple parameters that are not available for all molecules (RET, VET,...)
- A calibration is needed to determine the setup dependent constant

Absolute Calibration



- Known concentration (density) of the gas: NO
- A different species exhibiting a very similar pattern: O and H (Xe and Kr)
- Rayleigh/Raman scattering
- Absorption spectroscopy

Planar LIF (PLIF)



Two-Photon Laser-Induced Fluorescence (TALIF)



Laser Scattering Techniques

Laser Scattering (1)

hν

Scattering:

Vibrational Raman

 $\Delta v = -1$

- Thomson: elastic, free electrons
- Rayleigh: elastic, neutrals
- Raman: inelastic, molecular



Laser Scattering (2)



 $P_{\lambda} = f L P_i \Delta \Omega \cdot n \cdot \frac{\mathrm{d}\sigma}{\mathrm{d}\Omega} \cdot S_{\lambda}(\lambda).$



where f is a constant factor that takes into account the efficiency of the optics and camera, L the length of the detection volume along the laser path, P_i the incident laser power, $\Delta\Omega$ the solid angle of detection, *n* the density of the scattering particle, and $d\sigma/d\Omega$ the differential cross section. The factor $S_{\lambda}(\lambda)$ includes the spectral distribution as a function of wavelength,

Rayleigh Scattering



Van Gessel et al, Plasma Sources Sci. Technol. 21 (2012) 015003

- Rayleigh scattering cross sections are species dependent!
- Scatter off optics and dust particles produce strong unwanted signal

Laser Scattering Setup



Roettgen et al, Plasma Sources Sci. Technol. 25 (2016) 055009

• Triple grating spectrometer = filter to remove Rayleigh signal

Rotational Raman Spectrum



Thomson Scattering



wavelength (nm)

536

536

Time-resolved n_e , T_e and EEDF



Roettgen et al, Plasma Sources Sci. Technol. 25 (2016) 055009

Spatially-resolved Measurements: Microwave Plasma Jet



Thank you for your attention!!